3 g (54%) of XIII with mp 164-166°C. Mass spectrum: M⁺ 296.

LITERATURE CITED

REACTION OF 2-ALKOXYCARBONYLMETHYL DERIVATIVES OF Δ²-OXAZOLINE AND Δ²-IMIDAZOLINE AND THEIR TAUROMERS WITH 4-NITROBENZONITRILE N-OXIDE

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It was established that 2-alkoxy carbonylmethylene derivatives of oxazolidine and imidazolidine react readily with 4-nitrobenzonitrile N-oxide; the reaction takes place at the methyldiene carbon atom to give intermediate oximes, which can then undergo cyclization to isoxazoles. Their tautomers — benzimidazole and Δ²-oxazoline derivatives — react with considerably greater difficulty; in the first case the reaction takes place at a different center, viz., the ring nitrogen atom.

It is known that oxazolines and imidazolines in which a methylene fragment bonded to an electron-acceptor substituent (an ester group) is present in the 2 position can exist in two tautomeric forms [1-3]:

To study the effect of the tautomerism of derivatives I and II on their reactivities we investigated the reaction of these compounds with 4-nitrobenzonitrile N-oxide (III). The N-oxide was generated in situ by dehydrochlorination of 4-nitrobenzhydroxamic acid chloride with triethylamine. The starting I and II molecules contain several potential reaction centers, viz., an unsaturated bond (C= C or C=N), the ring nitrogen atom, and the methylene or methyldiene fragment. Since the reaction centers in tautomers I and II differ substantially, one might have expected that their reactions with nitrile oxide III would lead to different products. Nevertheless, we found that IV and V give derivatives of the same type in this reaction:

Oxazolidine IV, in which the structure is fixed and tautomerism is impossible, reacts very rapidly with nitrile oxide III, and the reaction is virtually complete immediately after mixing of the reagents at room temperature. In contrast to this, oxazoline V, for which tautomer Va predominates [2], reacts considerably more slowly. For the completion of the reaction 6-8 h are necessary. This difference in the reactivities can be explained by the fact that isoxazoline V reacts only in the tautomeric Vb form, the amount of which in the equilibrium mixture is small.

A similar pattern is observed in the imidazoline series. Derivative IX, which has a tautomeric form that is similar to oxazolidine IV, also reacts readily with nitrile oxide III:

However, in the case the reaction stops at the first step, and an acyclic addition product — oxime X — is formed. A product with a similar structure (VI) is evidently also an intermediate in the reactions of oxazolines.

The reaction of alkenes with nitrile oxides usually leads to cycloaddition products — isoxazolines [4]. The formation of acyclic products — unsaturated oximes — is observed extremely rarely, and only isolated examples of such reactions are known [5].

Under the influence of catalytic amounts of alkali oxime X readily undergoes cyclization to isoxazolone XI; in contrast to oxazoline derivatives, the heteroring is retained, and the ester group enters into the reaction.

According to PMR data, benzimidazole XII exists in only one tautomeric form similar to that which oxazoline Va has. This also evidently explains the fact that addition products similar to those obtained in the preceding cases are not formed in the reaction of XII with the nitrile oxide. The reaction proceeds very slowly and at a different reaction center, viz., the ring nitrogen atom:

Under the influence of alkali the ester group in benzimidazole XIII undergoes hydrolysis and decarboxylation to give 2-methyl derivative XIV, the structure of which was proved by alternative synthesis from 2-methylbenzimidazole and nitrile oxide III.

Thus, from the results obtained one may conclude that 2-methylene derivatives of imidazolidine and oxazolidine react readily with the nitrile oxide at the methylidyne fragment to give, in the first step, an acyclic addition product — an unsaturated oxime — which then can undergo cyclization to isoxazoles with the participation of the heteroring or the ester group. The $\Delta^2$-oxazoline reacts with the nitrile oxide with considerably greater difficulty but gives a similar product. The reaction evidently proceeds through a step involving tautomeric conversion of the $\Delta^2$-oxazoline to a 2-methyleneoxazolidine derivative. The benzimidazole derivative, for which this sort of tautomeric conversion is not observed, reacts differently — addition takes place at the ring nitrogen atom.

**EXPERIMENTAL**

The PMR spectra of solutions of the compounds in $d_6$-DMSO were recorded with a Bruker WH-90 spectrometer with tetramethylsilane (TMS) as the internal standard. The IR spectra of suspensions of the compounds in Nujol were recorded with a Perkin—Elmer 580B spectrometer. The course of the reactions and the purity of the compounds obtained were monitored by TLC on Silufol UV-254 plates with elution with ether and development in UV light.

The results of elementary analysis for C, H, and N were in agreement with the calculated values.

3-(4-Nitrophenyl)-4-ethoxycarbonyl-5-(2-hydroxethylamino)isoxazole (VIII, C$_{14}$H$_9$N$_3$O$_6$). A solution of 0.40 g (2 mmole) of 4-nitrobenzhydroxamic acid chloride in 10 ml of acetonitrile was added dropwise at room temperature to a solution of 0.31 g (2 mmole) of 2-ethoxycarbonylmethyl-$\Delta^2$-oxazoline and 0.22 g (2.2 mmole) of triethylamine in 15 ml of acetonitrile. After 6-8 h, the triethylamine hydrochloride was removed by filtration, the filtrate was